

Use of isotopes in metabolomics

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2-17-16

Synopsis

- **Natural abundance isotopes**
- **Tracing a metabolic pathway**
 - Labeling a precursor for qualitative analysis
 - 95% isotope/5% unlabeled and 5% unlabeled/95% isotope
- **Following individual carbon atoms**
- **Quantitative analysis of metabolic flux**
- **Post-extraction isotopic labeling**

Value of natural isotopes

- The natural abundance of isotopes enables the investigator to determine the charge state of an ion
 - The principal contribution to $[M+H]^+$ or $[M-H]^-$ isotope ions comes from ^{13}C (~1.1% of all carbon atoms)
 - The intensity of the ^{13}C isotope ion increases relative to the number of carbon atoms
 - There is often an observable $^{13}\text{C}_2$ isotope peak

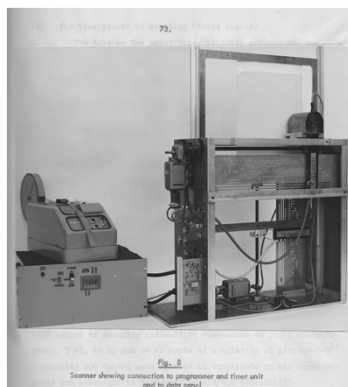
Value of the $[M+/-H+2]$ peak

- The mass difference due to a nominal increase in mass of 2 contains a lot of information
 - These are isotopic mass differences for each of the common elements
 - $^1\text{H}_2/{}^2\text{H}_2$ 2 x 1.006277 = 2.012554 (0.012%)
 - $^{12}\text{C}_2/{}^{13}\text{C}_2$ 2 x 1.003355 = 2.006710 (1.078%)
 - $^{14}\text{N}_2/{}^{15}\text{N}_2$ 2 x 0.997035 = 1.994079 (0.364%)
 - $^{16}\text{O}_2/{}^{17}\text{O}_2$ 2 x 1.004217 = 2.008434 (0.038%)
 - $^{16}\text{O}_2/{}^{18}\text{O}_1$ 1 x 2.004246 = 2.004246 (0.205%)
 - $^{32}\text{S}_2/{}^{33}\text{S}_2$ 2 x 0.999387 = 1.998774 (0.752%)
 - $^{32}\text{S}_2/{}^{34}\text{S}_1$ 1 x 1.995796 = 1.995796 (4.252%)

Using isotopes to trace a pathway

- Early studies (1930s) used ^2H , ^{13}C and ^{15}N labeling to map pathways
 - Limited to 1-200 m/z mass range
- 1950s/60s ^{14}C -radiotracers
 - 2D-Paper or thin layer chromatography
 - Radio gas chromatography
 - labeling of specific carbon atoms

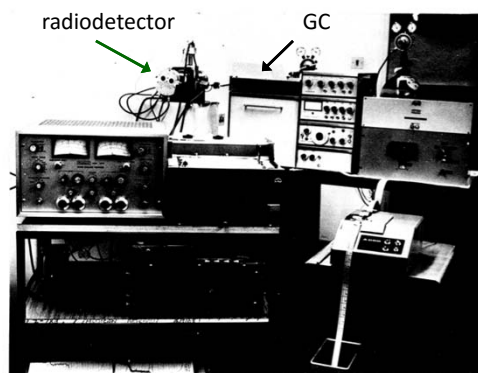
Origins of practical metabolomics Imperial College 1967-1970



Radio 2D-paper chromatography scanner with digitization of collected data

The room had 20 of these scanners – data analyzed by a central computer (in 1968)

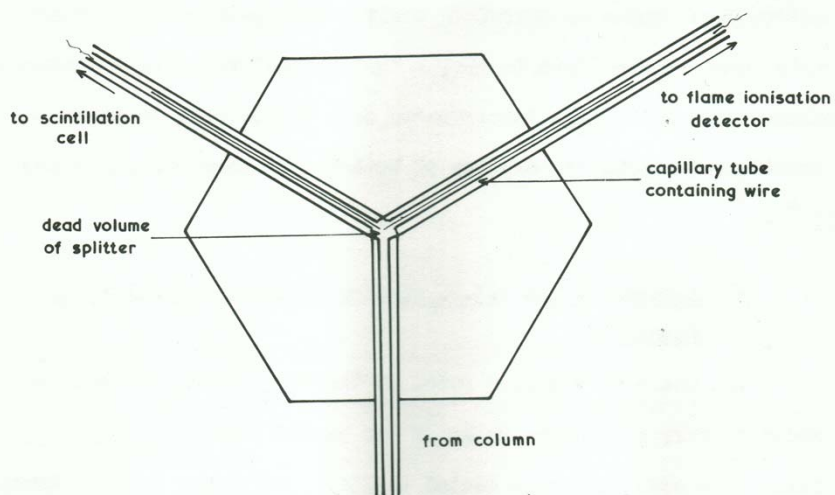
Courtesy of K.R. Mansford, PhD



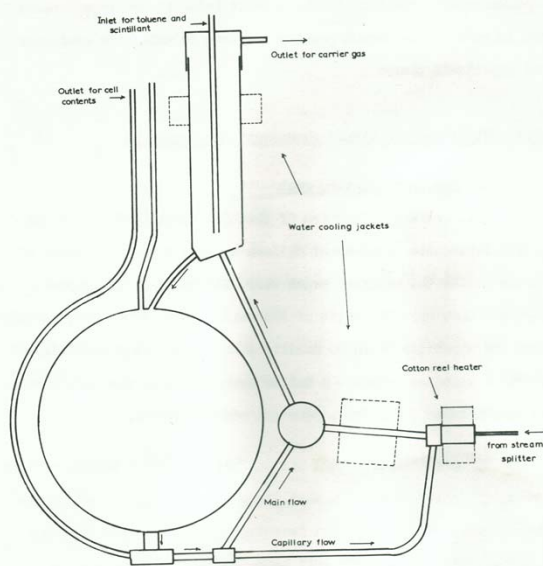
Radio gas-liquid chromatography with digitization of collected data

Developed this for my PhD work (1967-1970) to study glucose metabolism in acellular slime moulds

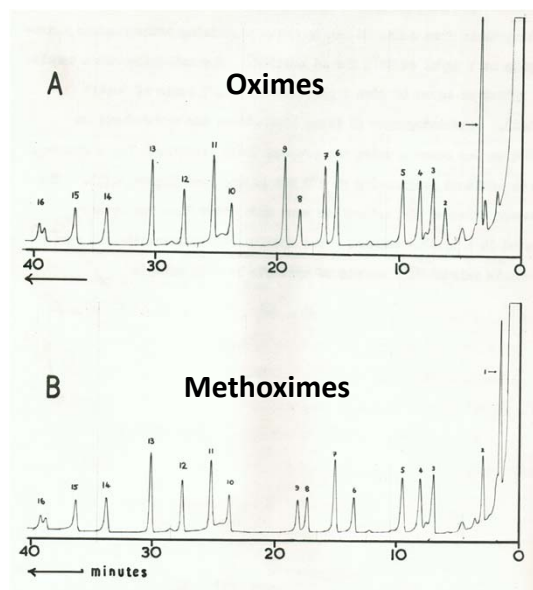
Stream splitter for radio GC



Popjak scintillation cell



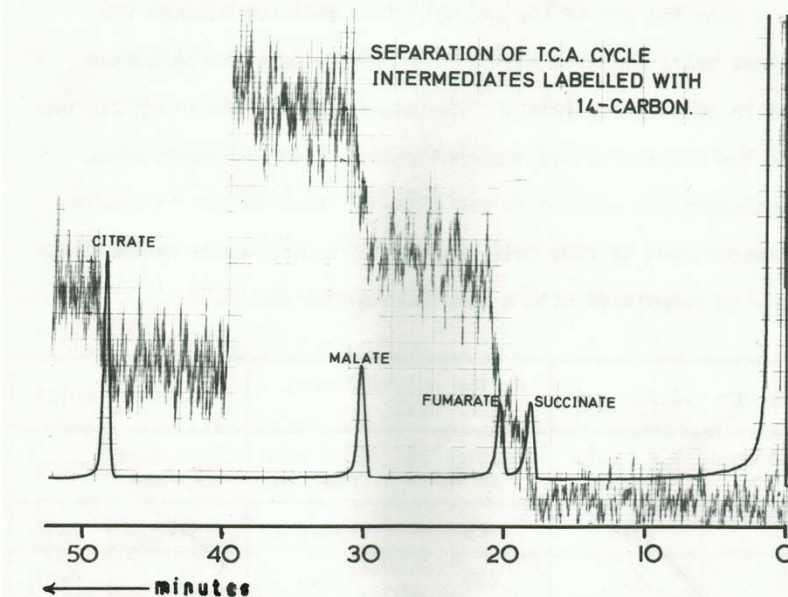
GC of glycolytic and Krebs cycle intermediates



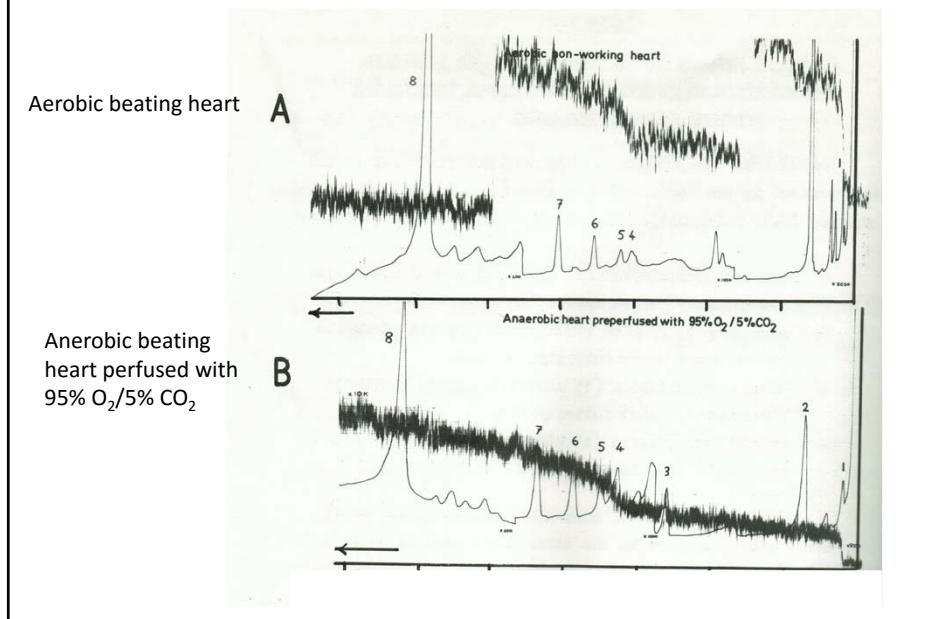
Temperature programming
of TMS ester/ethers on a 5' x
¼ inch packed column of
Chromosorb W coated with
OV-1 liquid phase

1= , 2= , 3= , 4= , 5= , 6= , 7= ,
8= , 9= , 10= , 11= , 12= , 13=
, 14= , 15= , 16=

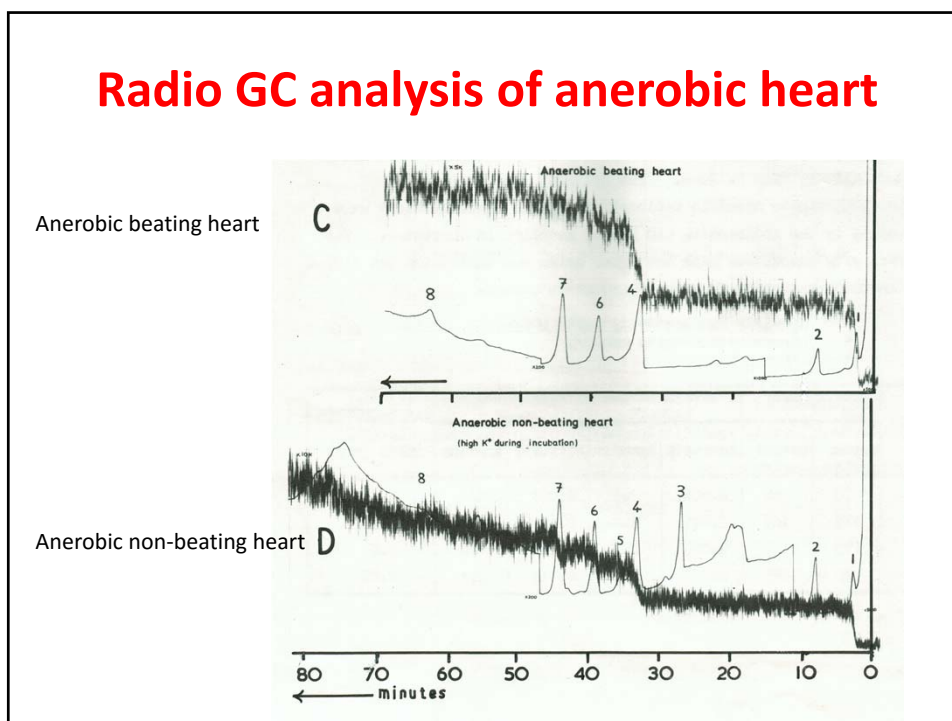
Radio-GC of Krebs Cycle intermediates



Radio GC analysis of beating heart



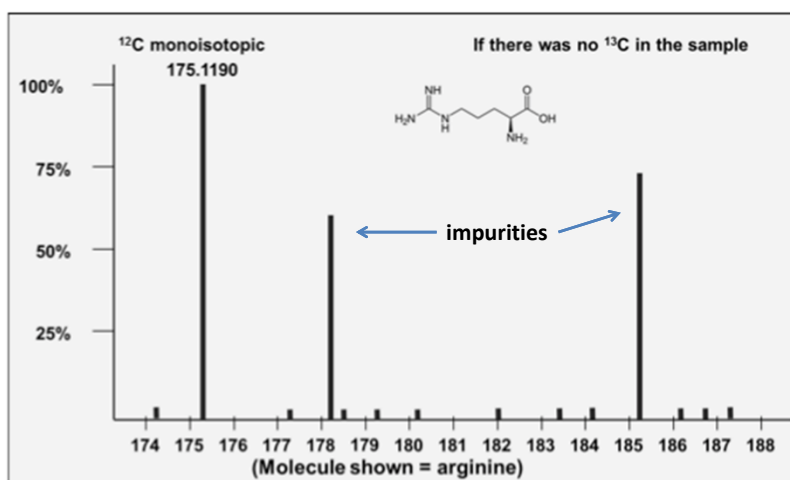
Radio GC analysis of anaerobic heart



Tracking metabolites with IROA

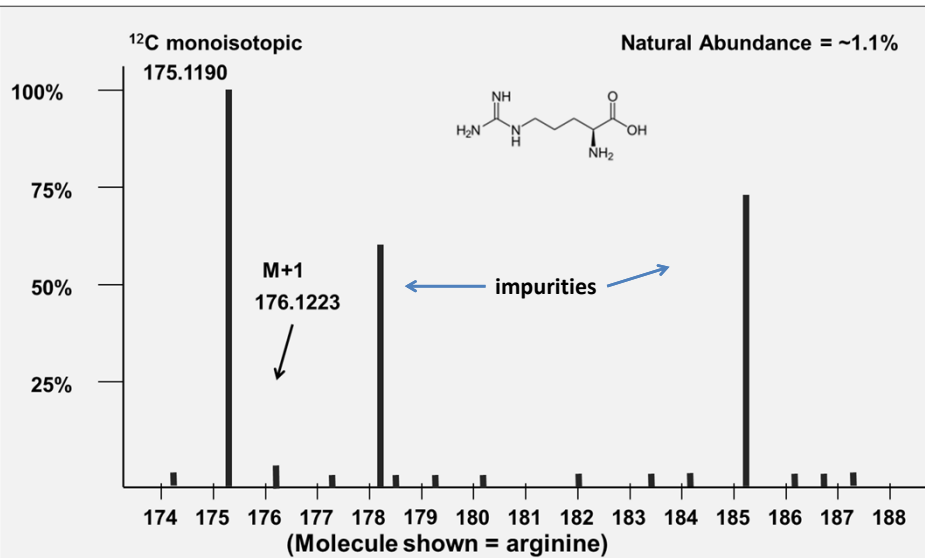
- Isotope ratio outlier analysis (IROA)
 - Not used for flux analysis, but rather to create a unique signal for metabolites
 - Used for LC-MS (and possibly GC-MS)
 - Designed to distinguish between metabolites of interest and background signals
 - Requires uniform labeling at the 95% and 5% ^{13}C -enrichment levels

All ^{12}C in arginine $[\text{M}+\text{H}]^+$



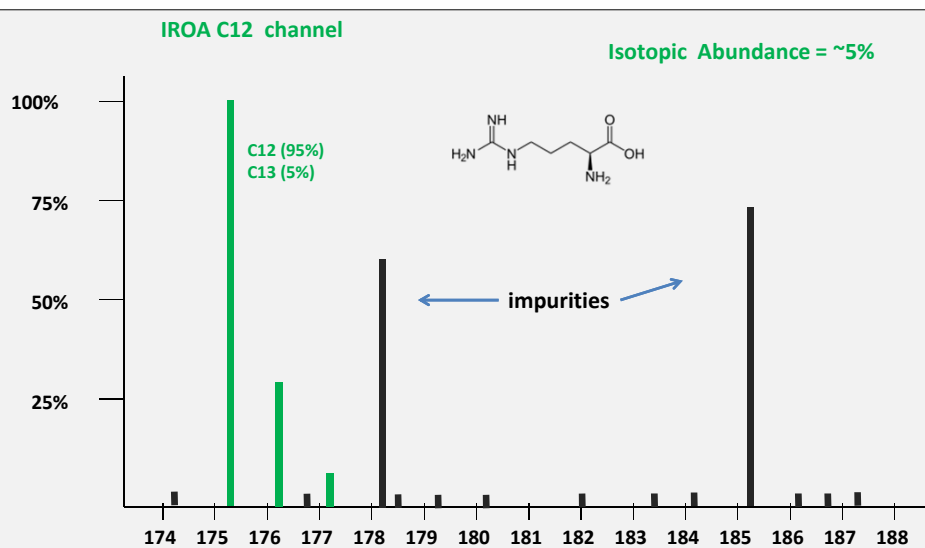
IROA Technologies

Natural abundance of ^{13}C in arginine



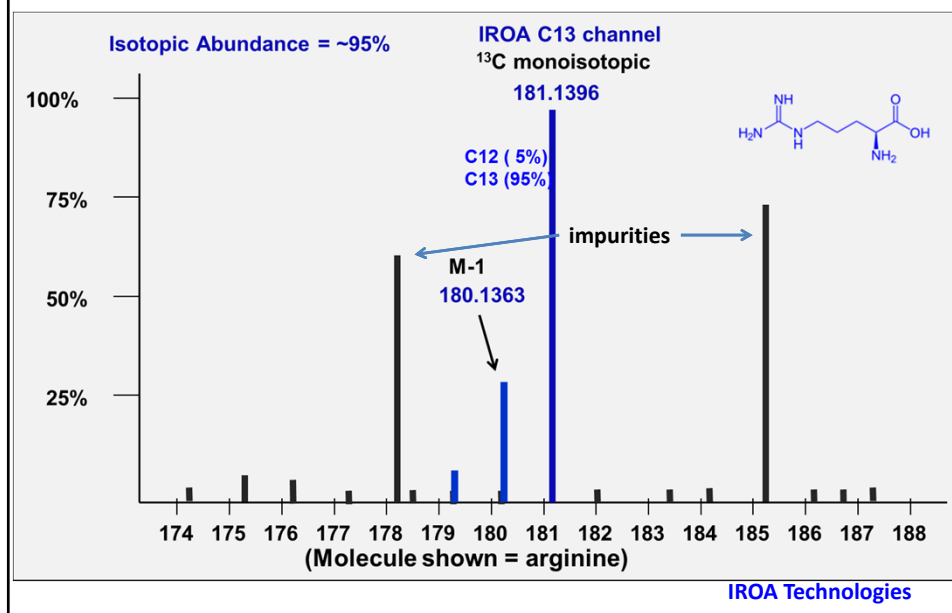
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Making ^{13}C abundance = 5%

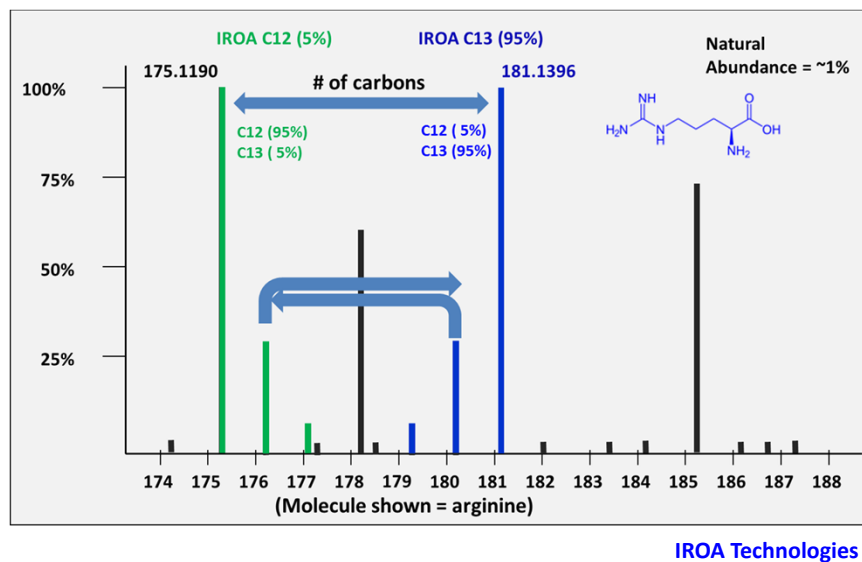


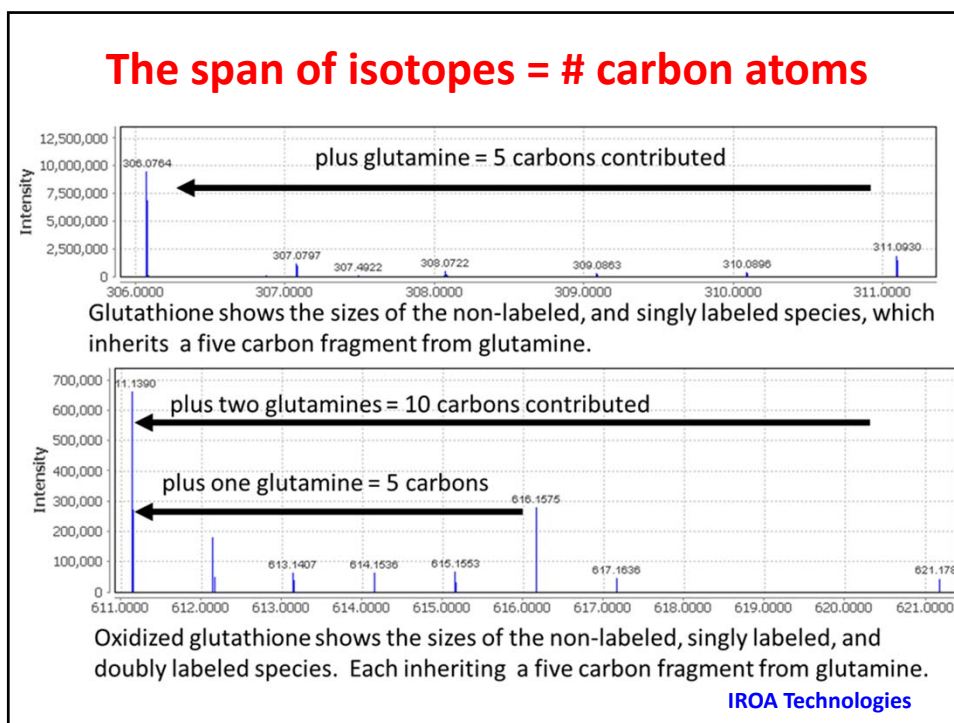
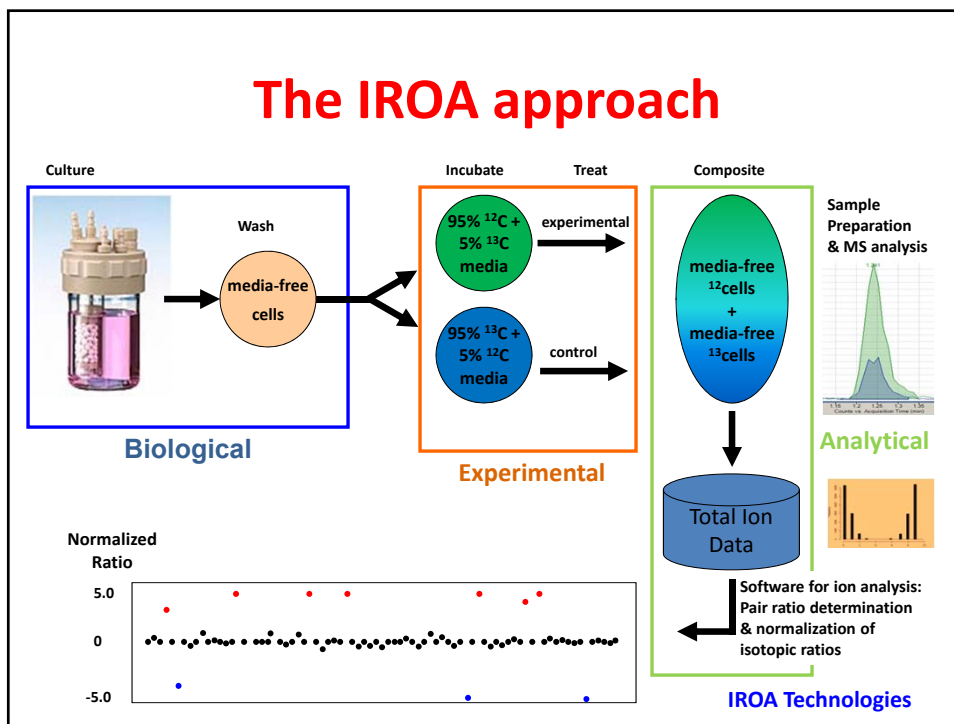
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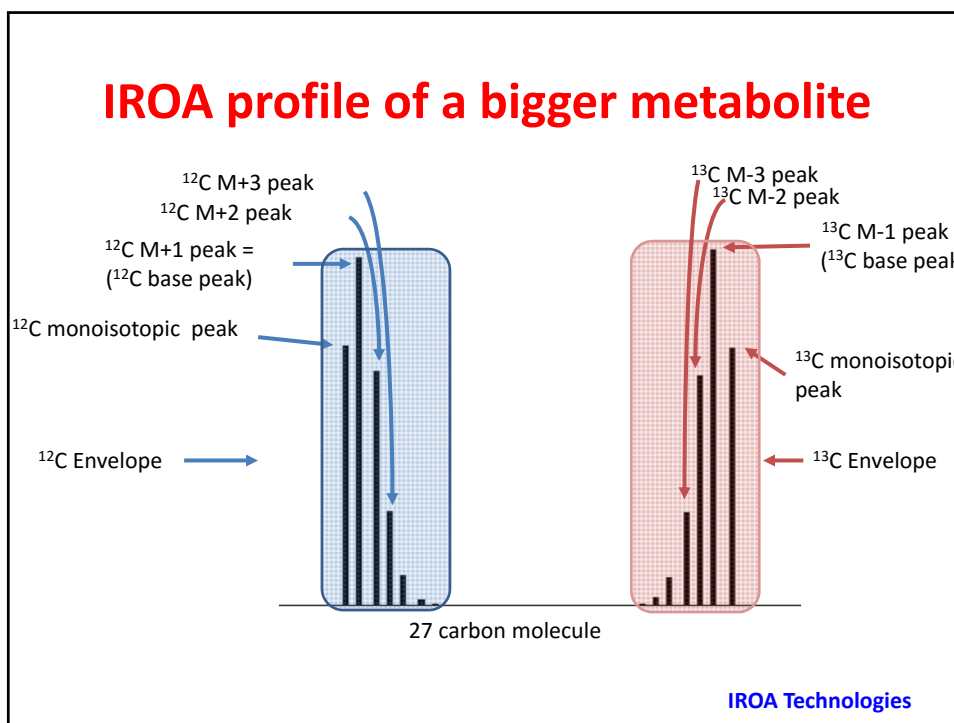
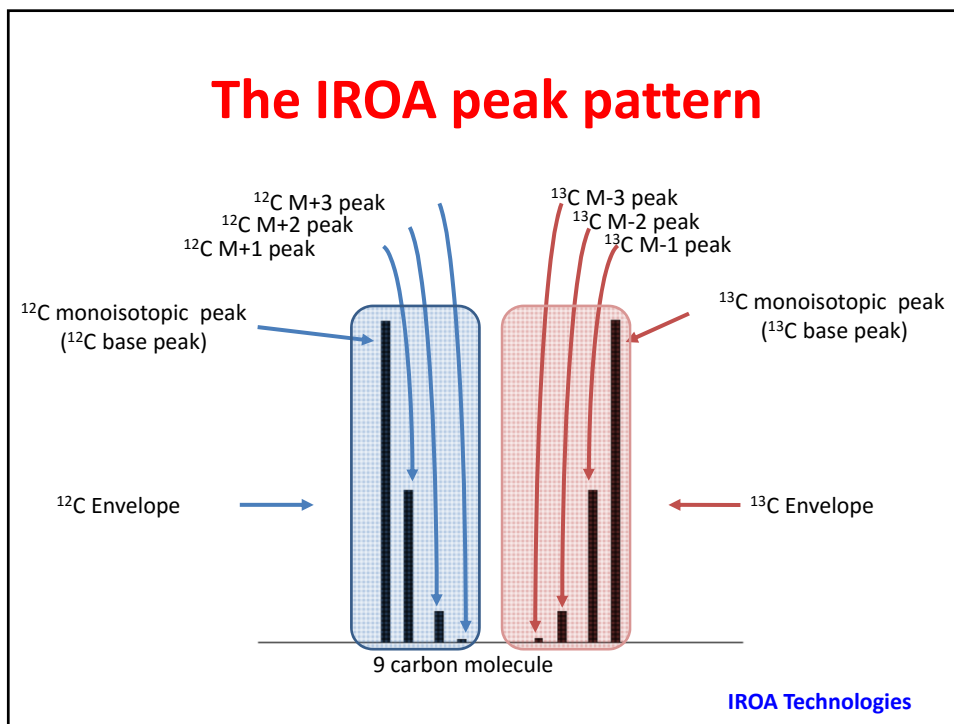
Making ^{13}C abundance = 95%

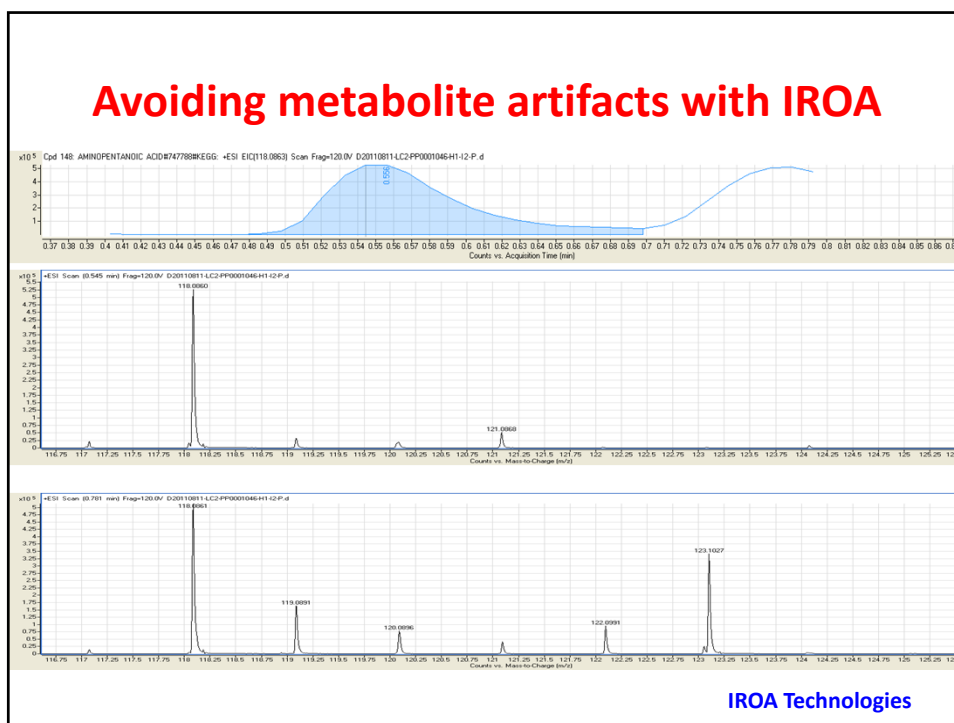
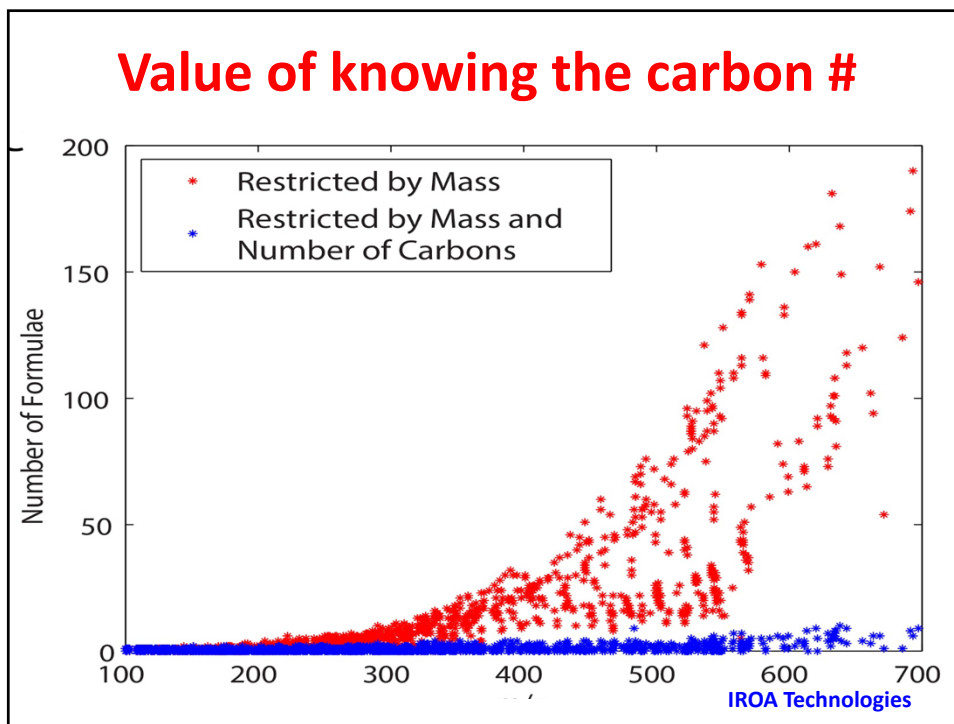


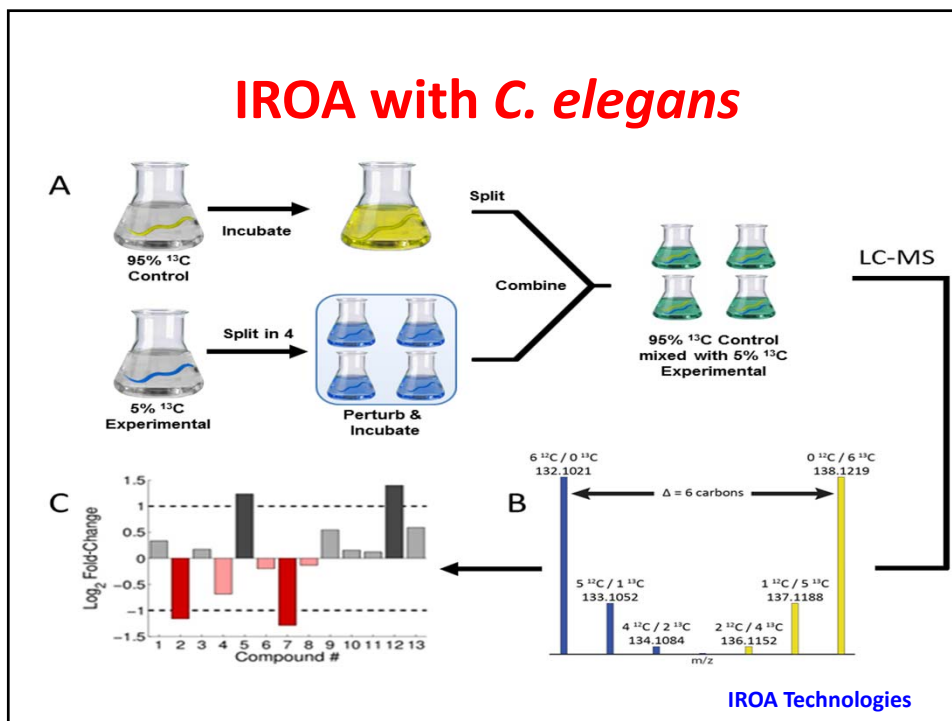
Pairing the 5% and 95% ^{13}C -labeling











Effect of a toxin on *C. elegans*

- **742 strong IROA peak pairs were found**
 - 314 named / 428 formula determined
 - Positive and negative mode LC
 - Thermo Orbi-trap @ 70K resolution
- **Strong response signature determined**
 - Basic statistics, PCA, Random Forest, NMF, SOM
 - 74 compounds were considered significant by at least 3 of these methods.

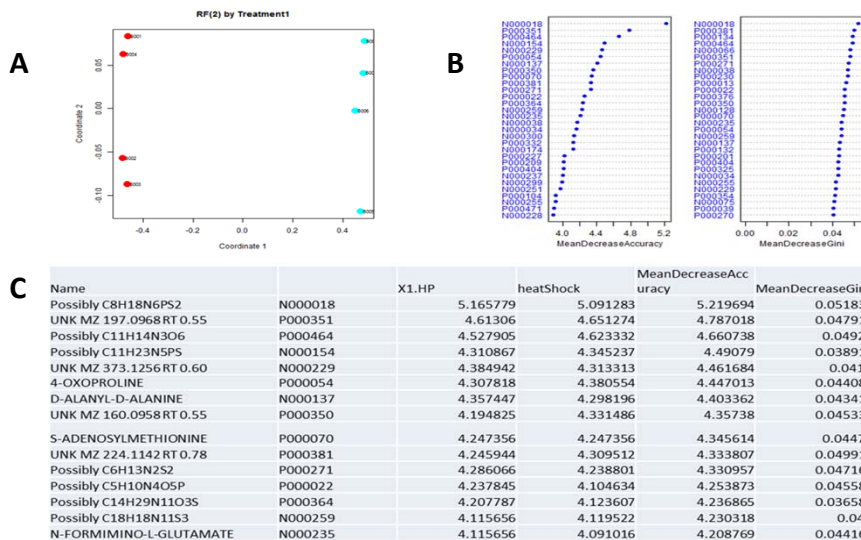
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Ions significantly affected by the toxin

Name		p.value	F-Value treatment ²
L-LYSINE	P000009	7.89E-05	89.71
Possibly C ₅ H ₇ N ₃ O ₅ S	P000018	3.06E-05	124.99
L-ARGININE	P000019	0.000131	74.84
Possibly C ₅ H ₉ NO ₁₁	P000025	0.000182	66.63
UNK m/z 369.2215 RT 0.58	P000040	2.19E-05	140.24
SACCHAROPINE	P000046	7.23E-05	92.51
L-THREONINE	P000051	2.64E-05	131.52
L-GLUTAMIC ACID	P000053	1.09E-06	389.79
4-OXOPROLINE	P000054	1.74E-05	151.81
Possibly C ₅ H ₇ NO	P000058	1.8E-05	150.26
L-VALINE	P000060	0.000262	58.37
CITRULLINE	P000061	3.15E-05	123.67
4-METHYLENE-L-GLUTAMINE	P000062	0.000169	68.40
L-METHIONINE S-OXIDE	P000065	7.55E-06	202.32
L-PROLINAMIDE	P000085	0.000227	61.56
STACHYDRINE	P000102	4.75E-05	107.19
UNK m/z 206.0368 RT 0.71	P000114	0.000251	59.35
N-ACETYLPUTRESCINE	P000122	8.96E-07	417.06
EPSILON-CAPROLACTAM	P000123	1.29E-08	1731.72
2-AMINO-OCTANOIC ACID	P000131	0.000213	62.99
UNK m/z 345.1258 RT 0.97	P000141	0.000111	79.36
Possibly C ₁₀ H ₁₉ N ₂ O ₅ P ₂	P000151	0.000154	70.78
CYS-GLY	P000152	0.000116	78.29
URATE	P000156	0.000222	62.02
Possibly C ₁₃ H ₁₆ N ₂ O ₅ P ₂	P000218	1.1E-05	177.82

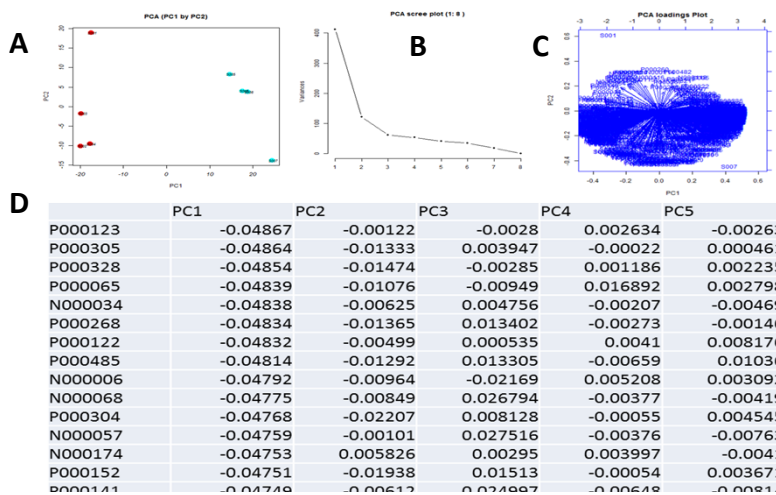
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Multivariate statistics



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PCA analysis of toxin's effect



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Summary of most likely metabolites

Name		Stats1	RFTop1	RFTop2	NMF3	NMF4	NMF5	NMF6	Count
UNK <i>m/z</i> 160.0958 RT 0.55	P000350	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 197.0968 RT 0.55	P000351	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 216.0852 RT 0.61	N000034	1	1	1	1	1	1	1	7
D-ALANYL-D-ALANINE	N000137	1	1	1	1	1	1	1	7
Possibly C ₂₅ H ₃₄ N ₄ O ₅	N000174	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 373.1256 RT 0.60	N000229	1	1	1	1	1	1	1	7
2-AMINO-OCTANOIC ACID	P000131	1	1	0	1	1	1	1	6
Possibly C ₈ H ₈ N ₄ O ₃	P000354	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 510.2122 RT 0.68	P000373	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 224.1142 RT 0.78	P000381	0	1	1	1	1	1	1	6
Possibly C ₆ H ₉ NO ₄ P	P000410	1	1	0	1	1	1	1	6
Possibly C ₁₁ H ₁₄ N ₄ O ₄	P000464	0	1	1	1	1	1	1	6
Possibly C ₆ H ₈ N ₄ O ₄ P	P000471	1	1	0	1	1	1	1	6
Possibly C ₈ H ₁₂ N ₄ O ₄ PS	N000006	1	1	0	1	1	1	1	6
Possibly C ₁₁ H ₂₃ N ₄ PS	N000154	1	1	1	0	1	1	1	6
D-GLUCOSE	N000228	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 548.2037 RT 0.63	N000232	1	1	0	1	1	1	1	6
GLYCERATE	N000237	1	1	0	1	1	1	1	6

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